

A Comparison of Different Soil Phosphorus Extraction Methods for Used to Determine Plant Available Soil Phosphorus of Erzurum Plain Agricultural Soils

Nesrin YILDIZ

Atatürk University, Faculty of Agriculture,
Dept.Of Soil Science and Plant Nutrition,
Erzurum, Turkey.

Tülay DİZİKISA

Agri Vocational Training School,
Ibrahim Çeçen University,
Agri, Turkey.

ABSTRACT

Phosphorus (P) fertilization is commonly based on soil testing, for which a variety of different soil P extraction methods are in use.

The purpose of this investigation, Phosphorus status of Erzurum plain (centre) soils was to determine and also to find out suitability of different extraction methods (distilled water, NaHCO₃, and sequential extraction method Step 1. and 6.) in determining of the plant available Phosphorus in Erzurum region soils. Representative 19 soils samples were collected from different soil locations. In order to select the most suitable chemical method as related to Phosphorus content of test plant (potato) was taken as the standart (biological) index. The results of the statistical analyses indicated that, all chemical extraction methods were not interrelated with biological indexes for the plant available phosphorus investigated in Erzurum plain soils.

Keywords: Availability ,Soil-P tests ,Extraction Methods, Plant uptake ,Correlation, P-Availability indexes

1. INTRODUCTION

A plant nutrient is a chemical element that is essential for plant growth and reproduction. Essential element is a term often used to identify a plant nutrient. The term nutrient implies essentiality, so it is redundant to call these elements essential nutrients. Seventeen elements are considered to have met the criteria for designation as plant nutrients. Carbon, hydrogen, and oxygen are derived from air or water. The other 14 are obtained from soil or nutrient solutions. For all the nutrients, their roles in agriculture were the subjects of careful investigations long before the elements were accepted as nutrients. One of them is Phosphorus (P). P is a central element to life on Earth. Living organisms are dependent on a persistent supply of P as it is crucially involved in most major metabolic processes, e.g. in energy transfer as adenosine triphosphate (ATP). Soil testing is a common approach to assessments of soil fertility and plant nutrition. With correlation to plant growth, development, and yield, soil testing indicates the capacity of soils to supply plant nutrients and

suggests appropriate corrective measures. Plant analysis, used in conjunction with plant symptoms and soil testing, is another common tool for assessment of the nutritional status of plants. As noted in a previous section, crop response to phosphorus is correlated poorly to the total amount of phosphorus in a soil. Therefore, a successful soil test should represent some index of phosphorus availability. The development of a soil test requires selection of an extractant, development of studies that correlate the amount of nutrient extracted with phosphorus accumulation by crops, and calibration studies that determine a relationship between soil test results and amount of fertilizer required for optimal production (Barker and Pilbeam, 2007).

Phosphorus (P): Mineral soils low in organic matter content, long history of cropping without adequate P fertilization reducing the supply of P, P-rich soils lost by erosion, calcareous soils where P availability is reduced by alkaline pH, the rate at which this process occurs depends on a number of soil factors: pH, soil moisture content, physiochemical characteristics of the colloidal substances, solubility characteristics of the solid-phase components, temperature biological activity (Jones, 2012).

The development of a soil test requires selection of an extractant, development of studies that correlate the amount of nutrient extracted with phosphorus accumulation by crops, and calibration studies that determine a relationship between soil test results and amount of fertilizer required for optimal production. Generally, water or dilute salt solutions characterize phosphorus in the soil solution or the intensity factor, whereas acids, complexing solutions, or alkaline buffer solutions generally characterize the quantity factor. Tests based on water extraction often correlate well with phosphorus accumulation in shallowrooted, fast-growing vegetable crops. However, soil tests capable of better characterizing the labile fraction and capacity factor generally produce more reliable results for field and orchard crops. As noted in the previous section, the amounts of phosphorus applied to crops should be based ideally on a well-calibrated soil test. However, even at a given soil-test phosphorus level, the amount of phosphorus fertilizer required for economic-optimum yield often will vary with crop. Generally, fast-growing, short-season vegetable crops have higher

A Comparison of Different Soil Phosphorus Extraction Methods for Used to Determine Plant Available Soil Phosphorus of Erzurum Plain Agricultural Soils

phosphorus requirements than field and orchard crops. Many deciduous fruit crops infrequently respond to phosphorus fertilization even if soil tests are low. It is presumed often that surface soil tests fail to characterize the full soil volume where trees take up nutrients or the fact that trees take up nutrients over a considerable time period (Barker and Pilbeam, 2007).

A chemical method for estimating the nutrient supplying capacity of a soil; measures a portion of a nutrient from a 'pool' that is used by plants; an index of nutrient availability; does not measure the total amount of nutrient in the soil; needs to be calibrated in field/greenhouse rate studies to then use in nutrient (fertilizer) recommendations. Can determine soil nutrient status before a crop (field, vegetable, ornamental) is planted (Carrie, 2008; Heckendorn, 2007)

Research on the selection of chemical extraction method has been done for different climate and will be continued for the future of all different soil and plant nutrient in Erzurum region (Yildiz and et al.1999; Yıldiz and et al.2003; Yıldiz and Güler 2010a; Yıldiz and Güler.2010b; Yıldiz and et al. 2010; Dizikisa and Yıldiz 2016a; Dizikisa and Yıldiz 2016a).

2. MATERIAL AND METHOD

Soils from 19 representative were sampled from potato grown fields in early April, 2010 with the aim of defining the nutrient potential in potato plants cultivated in central Erzurum. Soil samples from 0-40 cm depth in selected particular stations were taken and sieved with a 2mm mesh screen to analyse the different chemical properties and soil nutrient status. Leaf tissue was oven dried at 68 °C for 48 hours and ground to pass through a 1 mm mesh screen. The potato plant leaf sampled in start flowering from the 4th leaf plant leaf sample was taken June 2010 (Yildiz and Dizikisa, 2016).

3. RESULTS AND DISCUSSION

Plant phosphorus content of potato leaf were determined (Table1) (Yildiz and Dizikisa,2016).

Table.1 The results of Biological indexes and The concentrations of P Obtained distilled water, NaHCO₃, and sequential extraction methods (Step 1. and 6.)

Determining of available P contents of Erzurum plain soils 3 different chemical methods were used results shown in table1. The results of this study showed that plant available P obtained with distilled water, sodium bicarbonate, and sequentially extraction 1 and 6.step methods were interrelated with each other. But, the results of this study showed that plant available phosphorus obtained with distilled water, NaHCO₃, and sequential extraction step 1. and 6. methods (Table 1) were not interrelated with phosphorus content of potato leaf (biological index) in Erzurum Plain soils (Table 2). Results also showed that the 3 different chemical extraction methods might not be used for plant available P at least in this conditions growing potato in this location.

Before deciding that the methods are completely unsuitable in evaluating plant available P under different plants, climate and soil conditions the methods have been tried extensively in other experimental conditions, greenhouse and field etc.

Although application of fertilizer in irrigation water (fertigation) is a common practice with mobile nutrients such as nitrogen, it is less common with phosphorus because of concerns about efficiency of utilization. Owing to the soil reactions discussed in a previous section, it is often presumed that much of the phosphorus applied with water will be tied up at its point of contact with the soil.

Acknowledgements

We would like to thank the Ataturk University a part for funding the project (BAP. 2011/194)

REFERENCES

- 1) **Baker, A. V. and Pilbeam, D. J., 2007.** Handbook of Plant Nutrition. CRC Press Taylor & Francis Group 6000 Broken Sound Parkway NW, Suite 300 Boca Raton, FL 33487-2742.
- 2) **J. Benton Jones, Jr., 2012.** Plant Nutrition and Soil Fertility Manual, Second Edition. CRC Press. Taylor & Francis Group.6000 Broken Sound Parkway NW, Suite 300. Boca Raton, FL 33487-2742 by Taylor & Francis Group, LLC. CRC. Press is an imprint of Taylor & Francis Group, an Informa business
- 3) **Carrie, L., 2008.** Soil testing and Plant analysis. Dept. Of Soil Science. U.V-Madison.
- 4) **Cheng,Yuan, C.,Hung-Lung, C.,Zhao-Ji, S. and Chu-Fang, W., 2005.** A sequential extraction method measures the toxic metal content in fly ash from a municipal solid waste incinerator. Journal of the Chinese Chemical Society, 52, 921-926
- 5) **Dizikisa, T., Yıldız N., 2016a.** Estimation Of Nutritional Status Of Potato Solanum Tuberosum L Plant By Soil And Leaf Analyses Grown In Pasinler Town Plain of Erzurum. EI: Engineering Index International Journal of Innovative Research in Engineering& Management (IJIREM) 6, 2349-2058 234 -238
- 6) **Dizikisa, T., Yıldız N., 2016b.** Estimation of Nutritional Status of Potato Solanum Tuberosum L Plant by Soil and Leaf Analyses Grown in Oltu Town of Erzurum EI: Engineering Index International Journal of Innovative Research in Engineering& Management (IJIREM) 6, 2349-2058, 234- 238
- 7) **Forsee Jr. W.T. 1942.** Development and evaluation of methods for the determination of phosphorus in Everglades peat under various conditions of treatment. Soil Sci. Soc. Fla. Proc. 4:50-54, 1942.
- 8) **Heckendorn, S., 2007.** Soil Testing and Plant Analysis. Virginia Cooperative Extension.
- 9) **Jackson, M. L., 1962.** Soil Chemical Analysis. Prentice-Hall. Inc. Cliffs, USA
- 10) **Olsen, S.R. , Cole, C.V., Watanabe F.S., and Dean L.A., 1954.** Estimation of Available P in Soils by Extraction with NaHCO₃. USDA Cir. 939. US Government Printing Office, Washington DC,
- 11) **Tessier, A., Campbell, P. and Bisson, M., 1979.** Sequential extraction procedure for the speciation of particulate trace metals. Analytical Chemistry, 51: 844-851.
- 12) **Yıldız, N., Canbolat M. Y. ve Aydemir O., 1999.** Erzurum Daphan Ovası Topraklarının Bitkiye yararlı Azot Durumunun Değerlendirilmesi. GAP. I. Tarım Kongresi. 1043-1049, Şanlıurfa.
- 13) **Yıldız, N., Güler, E., Bilgin, N., Kahraman, F., Akkuş, F., Er, G. ve Diyarbakırlı, S., 2010.** Erzurum Ovası Topraklarının Kalsiyum, Magnezyum ve Molibden Durumunun Neubauer Fide Yöntemi ile Belirlenmesi. 5. Bitki Besleme ve Gübre Kongresi. 15-17 Eylül. Ege Üniversitesi. Ziraat Fakültesi Dergisi. Özel Sayı. ISSN: 1018-8851., 447-452, İzmir.
- 14) **Yıldız, N ve Güler E., 2010a.** Erzurum Ovası Tarım Topraklarının Bitkiye Yararlı Bor Durumunun Uygun Ekstraksiyon Yöntemleri Seçilerek Değerlendirilmesi. 5. Bitki Besleme ve Gübre Kongresi. 15-17 Eylül. Ege Üniversitesi.

Ziraat Fakültesi Dergisi. Özel Sayı. ISSN: 1018-8851., 458-464. İzmir.

15)Yıldız, N ve Güler E., 2010b. Erzurum Ovası Tarım Topraklarının Bitkiye Yararışlı Bor Durumunun Uygun Ekstraksiyon Yöntemleri Seçilerek Değerlendirilmesi. 5. Bitki Besleme ve Gübre Kongresi. 15-17 Eylül. Ege Üniversitesi. Ziraat Fakültesi Dergisi. Özel Sayı. ISSN: 1018-8851., 458-464. İzmir.

16)Yıldız, N., Bilgin N. ve Aksu E., 2003. Erzurum-Daphan Ovası topraklarının fosfor durumunun değerlendirilmesi. GAP.III. Tarım Kongresi. 2-3 Ekim, 583-587, Şanlıurfa.

17)Yıldız, N., and Dizikusa, T., 2016. Estimation Of Nutritional Status Of Potato Solanum Tuberosum L Plant By Soil And Leaf Analyses Grown In Erzurum Center EI: Engineering Index International Journal of Innovative Research in Engineering & Management (IJIREM) , 3/2350-0557 241- 245.

18)Yıldız, N., Bilgin, N. ve Barik, K., 2016. Determination of plant available nitrogen of Erzurum plain soils SCI-Expanded Asian Journal of Chemistry 4, 3049 -3056.

Table.1 The results of Biological indexes and The concentrations of P Obtained distilled water, NaHCO₃, and sequential extraction methods (Step 1. and 6.)

P (ppm)					Plant P content (%)
Sequential extraction method			Water soluble P	Olsen (NaHCO ₃)	
Samp. No	Step.1	Step.6			
1	<0.00	12851	10,91	27	0,15
2	<0.00	3129	4,98	14	0,15
3	<0.00	9160	4,40	15	0,14
4	2,53	9315	18,04	45	0,18
5	<0.00	6234	4,73	21	0,22
6	0,78	7262	5,68	16	0,18
7	1,88	12394	9,41	33	0,21
8	23,22	27319	31,60	65	0,23
9	6,46	12375	21,92	45	0,20
10	0,11	5144	12,88	35	0,21
11	1,04	16121	10,10	31	0,22
12	0,13	5136	1,21	06	0,25
13	2,73	18165	3,28	19	0,19
14	2,25	2879	5,40	21	0,19
15	<0.00	4497	3,42	09	0,17
16	<0.00	4207	3,80	20	0,16
17	0,07	3353	8,87	18	0,16
18	0,04	14393	6,76	24	0,14
19	1,08	4694	6,63	31	0,17

A Comparison of Different Soil Phosphorus Extraction Methods for Used to Determine Plant Available Soil Phosphorus of Erzurum Plain Agricultural Soils

Table.2 The linear correlation coefficients between chemical extractable P concentrations (ppm) with plant P % (Biological index) for Erzurum Plain Soils

	S.E (step 1)	S.E(step 6)	Water	NaHCO ₃	Plant P.cont (%)
S.E (step 1)	1				
S.E (step 6)	0,729**	1			
Water	0,825**	0,633**	1		
NaHCO ₃	0,770**	0,651**	0,944**	1	
Plant P.cont (%)	0,395	0,270	0,290	0,307	1

**; Significant at the 0,01 probability

Water soluble; distilled water; 5 g soil/ 50 ml dist. water (Forse,1942)

Plant leaf P content (%); P content of Potato leaf Biological index (Yildiz and Dizikisa, 2016)

NaHCO₃ ; 0,5 M NaHCO₃ , pH;8,5 (Olsen, 1954)

S.E; Sequential Extraction (step1 and 6) Procedure; Water soluble P, step1; 0,5 g soil/10 ml distile water step 6; For silica bound P, residual soil / HNO₃+H₂O₂+ HF (3;5;2 ml) high -pressure bomb digestion at 190 °C (Tessiar 1979 ; Cheng et al 2005)).